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of a barometer.
standard temperature
& pressure (STP)
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law of partial
pressures. Definitions
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CHAPTER 11 REVIEW

Gases Class SHORT

ANSWER Answer the following questions in the space provided.

Increasing rate of effusion. List the following gases in order (Assume all gases are at the same temperature and pressure.) (c) HCl (d) 2. Explain your reasoning for the order of gases

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you chose in item 1 above. Refer to the kinetic-

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This might also be a good time to read the Chapter Objectives, which precede the Review Questions. Chemists try to “see” the structure of matter even more closely than can be seen in any photograph. ... negative charge.

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Chemistry**

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Atomic Theory Review

Skills 11.1 The

Mysterious Electron

Standing Waves and

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Guitar Strings
Electrons as Standing Waves
Waveforms for Hydrogen Atoms
Particle Interpretation of the Wave Character of the Electron
Other Important Waveforms
Overall Organization of Principal

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true or false. Students
relate information
learned about
introductory knowledge
gained in chemistry to
accurately answer the
given questions.

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CHAPTER 11 REVIEW
Gases SECTION 2
SHORT ANSWER

Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the

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following
circumstances:

- increase a.
temperature increases,
volume stays the same
decrease b. volume
increases, temperature
stays the same

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